

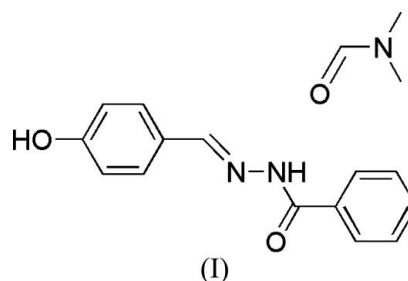
Zuo-Liang Jing, Zhi Fan,\* Ming  
Yu, Xin Chen and Qi-Liang DengCollege of Sciences, Tianjin University of  
Science and Technology, Tianjin 300222,  
People's Republic of China

Correspondence e-mail: zhifan@public.tpt.tj.cn

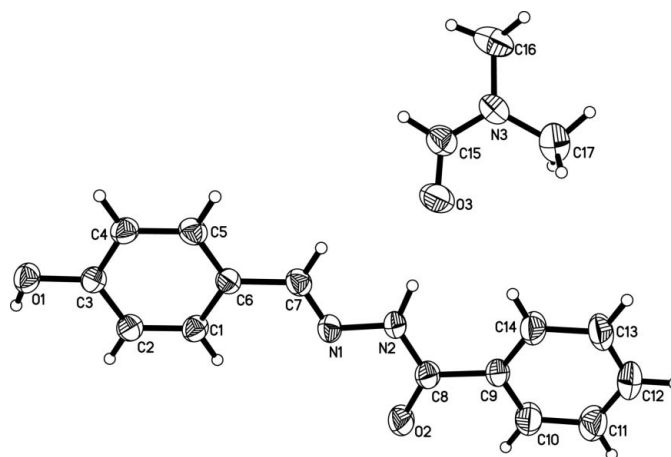
## Key indicators

Single-crystal X-ray study  
 $T = 294$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å  
 $R$  factor = 0.046  
 $wR$  factor = 0.149  
Data-to-parameter ratio = 15.7For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.***N'*-(4-Hydroxybenzylidene)benzohydrazide  
dimethylformamide solvate**The components of the title solvate structure,  $\text{C}_{14}\text{H}_{12}\text{N}_2\text{O}_2 \cdot$   
 $\text{C}_3\text{H}_7\text{NO}$ , are connected *via*  $\text{N}-\text{H} \cdots \text{O}$ ,  $\text{O}-\text{H} \cdots \text{N}$  and  $\text{O}-$   
 $\text{H} \cdots \text{O}$  hydrogen bonds, leading to a chain motif.Received 23 August 2005  
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## Comment

One of the aims of investigating the structural chemistry of  
Schiff bases is to develop protein and enzyme mimics (Santos  
*et al.*, 2001). As part of an investigation of the coordination  
properties of Schiff bases functioning as ligands, the present  
study details the crystal structure of the title compound, (I).

The asymmetric unit of (I) comprises one molecule each of *N'*-(4-hydroxybenzylidene)benzohydrazide and dimethylformamide (Fig. 1). The key  $\text{N1}-\text{N2}$ ,  $\text{N1}=\text{C7}$  and  $\text{C6}-\text{C7}$  bond lengths are 1.384 (3), 1.274 (3) and 1.465 (3) Å, respectively, which are consistent with those found in 2-hydroxy-3-methoxybenzaldehyde 2,4-dinitrophenylhydrazone (Jing *et al.*, 2005). The non-H atoms of the phenol group are coplanar, with an r.m.s. deviation of 0.017 Å, and similarly the r.m.s. deviation of the  $\text{N1}/\text{N2}/\text{C8}-\text{C14}/\text{O2}$  atoms from their least-squares plane is 0.073 Å. The dihedral angle between the



**Figure 1**  
The structure of (I), with displacement ellipsoids drawn at the 30% probability level.

aformentioned planes is  $8.17 (10)^\circ$ . Intermolecular N—H...O, O—H...N and O—H...O hydrogen bonds stabilize the crystal packing (Fig. 2). As detailed in Table 1, the H atom on O1 forms a strong interaction with atom O2<sup>i</sup> and these extend along the *b* axis to form a chain [symmetry code: (i)  $2 - x, y - \frac{1}{2}, \frac{1}{2} - z$ ]. Additional stabilization of the chain is afforded by somewhat weaker O1—H...N2<sup>i</sup> interactions. In this sense, the hydroxyl H atom might be thought of as being bifurcated. The dimethylformamide molecules are connected to this chain *via* N—H...O3 interactions.

## Experimental

An anhydrous ethanol solution of 4-hydroxybenzaldehyde (1.22 g, 10 mmol) was added to an anhydrous ethanol solution of benzohydrazide (1.36 g, 10 mmol) and the mixture was stirred at 350 K for 5 h under nitrogen, whereupon a yellow precipitate appeared. The product was isolated, recrystallized from ethanol and then dried *in vacuo* to give pure compound (I) in 87% yield; m.p. 504 K, literature 502 K (Kaupp *et al.*, 2000). Bright-yellow single crystals of (I) suitable for X-ray analysis were obtained by slow evaporation of a dimethylformamide solution.

### Crystal data

$C_{14}H_{12}N_2O_2 \cdot C_3H_7NO$	Mo $K\alpha$ radiation
$M_r = 313.35$	Cell parameters from 2528 reflections
Orthorhombic, <i>Pbca</i>	$\theta = 2.7\text{--}22.5^\circ$
$a = 15.518 (3) \text{ \AA}$	$\mu = 0.09 \text{ mm}^{-1}$
$b = 9.3215 (18) \text{ \AA}$	$T = 294 (2) \text{ K}$
$c = 22.942 (5) \text{ \AA}$	Block, yellow
$V = 3318.4 (11) \text{ \AA}^3$	$0.30 \times 0.22 \times 0.20 \text{ mm}$
$Z = 8$	
$D_x = 1.254 \text{ Mg m}^{-3}$	

### Data collection

Bruker SMART CCD area-detector diffractometer	3424 independent reflections
$\varphi$ and $\omega$ scans	1576 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan <i>SADABS</i> (Sheldrick, 1997)	$R_{\text{int}} = 0.081$
$T_{\text{min}} = 0.932, T_{\text{max}} = 0.983$	$\theta_{\text{max}} = 26.5^\circ$
17730 measured reflections	$h = -19 \rightarrow 14$
	$k = -11 \rightarrow 11$
	$l = -19 \rightarrow 28$

### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0552P)^2 + 0.9554P]$
$R[F^2 > 2\sigma(F^2)] = 0.046$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.149$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 0.98$	$\Delta\rho_{\text{max}} = 0.17 \text{ e \AA}^{-3}$
3424 reflections	$\Delta\rho_{\text{min}} = -0.15 \text{ e \AA}^{-3}$
218 parameters	Extinction correction: <i>SHELXL97</i>
H atoms treated by a mixture of independent and constrained refinement	Extinction coefficient: 0.0099 (9)

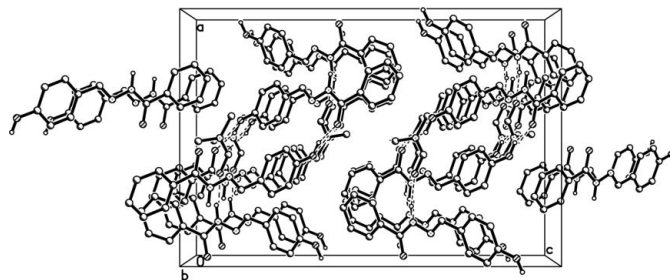


Figure 2

A view of the crystal packing in (I); hydrogen-bonding interactions are shown as dashed lines.

Table 1

Hydrogen-bond geometry ( $\text{\AA}, ^\circ$ ).

$D\text{--}H\cdots A$	$D\text{--}H$	$H\cdots A$	$D\cdots A$	$D\text{--}H\cdots A$
O1—H1...N1 <sup>i</sup>	0.91 (3)	2.64 (4)	3.316 (3)	132 (3)
O1—H1...O2 <sup>i</sup>	0.91 (3)	1.81 (4)	2.677 (3)	158 (3)
N2—H2...O3	0.97 (3)	1.91 (3)	2.864 (3)	166 (2)

Symmetry code: (i)  $-x + 2, y - \frac{1}{2}, -z + \frac{1}{2}$ .

C-bound H atoms were included in the riding-model approximation, with C—H bond lengths of 0.93 (aromatic) and 0.96  $\text{\AA}$  (methyl), and with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  and  $1.5U_{\text{eq}}(\text{C})$ , respectively. The O- and N-bound H atoms were refined freely (see Table 1).

Data collection: *SMART* (Bruker, 1999); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1999); software used to prepare material for publication: *SHELXTL*.

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## References

- Bruker (1999). *SMART, SAINT and SHELXTL*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Jing, Z.-L., Yu, M., Chen, X., Diao, C.-H., Deng, Q.-L. & Fan, Z. (2005). *Acta Cryst.* **E61**, o145–o146.
- Kaupp, G., Schmeyer, J. & Boy, J. (2000). *J. Prakt. Chem.* **342**, 269–280.
- Santos, M. L. P., Bagatin, I. A., Pereira, E. M. & Ferreira, A. M. D. C. (2001). *J. Chem. Soc. Dalton Trans.* pp. 838–844.
- Sheldrick G. M. (1997). *SADABS, SHELXS97 and SHELXL97*. University of Göttingen, Germany.